

1 **Role of magma mixing in the pre-eruptive dynamics of the Aeolian**
2 **Islands volcanoes (Southern Tyrrhenian Sea, Italy)**

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30 **Abstract**

31 We combined literature and experimental data to determine the role of magma mixing in the pre-
32 eruptive dynamics of the Aeolian Islands volcanoes. As a first step, we systematically reviewed the
33 evidence supporting the hypothesis of mixing-triggered eruptions in the Aeolian archipelago,
34 providing textural, chemical, and rheological constraints. The existing data highlighted the
35 significant role of magma mixing in many eruptions within the Aeolian archipelago. Examples
36 include the Upper Pollara and Porri volcano eruptions at Salina, Monte Guardia, and the AD 1230
37 Monte Pilato eruption at Lipari, as well as the present-day activity at Stromboli. Then, we focused
38 on Vulcano Island, chosen as a case study because it represents one of the volcanoes posing the
39 highest risk in the Aeolian archipelago. At Vulcano Island, we highlighted the role of magma
40 mixing in the AD 1739 and 1888–90 eruptions. Finally, we investigated mixing-to-eruption
41 timescales for the AD 1739 eruption, performing mixing experiments, and evaluated the progressive
42 decay of the chemical concentration variance with time. Results pointed to mixing-to-eruption
43 timescales of the order of 29 ± 9 h and magma ascent rates ranging between 3×10^{-2} and 5×10^{-2} m s⁻¹.
44 We finally emphasized that the presented results may have significant implications in the context of
45 volcanic hazard mitigation and planning of emergency activities.

46

47 **Key words:** magma mixing, time series experiments, eruption timescales

48 **1. Introduction**

49 The physical and chemical interaction between compositionally different magmas (i.e.
50 magma mixing in the broadest interpretation that includes mingling and mixing, in agreement with
51 Perugini and Poli, 2012) is a widely studied petrological phenomenon (e.g., Anderson, 1976; Sparks
52 et al., 1977; Bateman, 1995; De Rosa et al. 1996; Wiebe, 1994; De Campos et al., 2008; Albert et
53 al., 2015; Wiesmaier et al., 2015) that has been investigated considering either chemical (e.g.,
54 Leshner, 1990; Bateman, 1995; Bergantz et al., 2015; Perugini and Poli, 2005; Petrelli et al., 2006,
55 2011), thermo-dynamical (e.g., Sparks and Marshall, 1986; Snyder, 2000), or rheological
56 approaches (Blake and Fink, 2000; Albert et al., 2015). Moreover, magma mixing has been
57 suggested as an effective eruption trigger (e.g., Sparks et al. 1977; Druitt et al. 2012), and recently,
58 the combination of experimental investigations and the study of volcanic rocks produced by the
59 interaction between different magmas has been proposed as a robust volcanic chronometer (e.g.,
60 Perugini et al., 2015). The main idea behind the use of magma mixing as a volcanic chronometer
61 assumes the injection of hotter, more primitive, and potentially volatile-rich magma into a more
62 evolved magmatic reservoir as among the leading causes of mechanical destabilization of the
63 system, possibly triggering the eruption (e.g., Sparks et al., 1977; Eichelberger, 1980; Huppert et
64 al., 1984; Wiebe, 1994; Laumonier et al., 2014a, 2014b; Perugini et al., 2015).

65 Observation of pyroclastic rocks and lava flows strongly supports the hypothesis that
66 magma mixing is often involved in the volcanic plumbing system (i.e. the structural framework of
67 the pathways and storage regions through which magma travels from its source region to the Earth's
68 surface; Galland et al., 2015) before an eruption. Examples of significant eruptions from the last
69 two centuries providing evidence of magma mixing are Tambora (1815; Gertisser et al., 2011),
70 Mount St. Helens (1980; Gardner et al., 1995), Pinatubo (1991; Pallister et al., 1992), Soufriere
71 Hills (1995; Murphy et al., 1998), and Ejafjallajökull (2010; Sigmundsson et al., 2010). However,
72 even if the occurrence of a direct link between the injection of new magmas in a shallow crust

73 magmatic reservoir and a subsequent eruption was already established, the timescales of the process
74 remain unclear and warrant further investigation.

75 Substantial evidence of magma mixing has been found in many pyroclastic and effusive
76 products derived from the eruptive activity of the Aeolian Islands volcanoes (Southern Tyrrhenian
77 Sea, Italy), making this area of great interest to study the interaction between compositionally
78 different magmas prior to the eruptions (Francalanci et al., 1989, 1999; Peccerillo, 2005; Petrone et
79 al., 2006; 2018; De Rosa et al., 1996, 2002).

80 In this paper, we report a comprehensive review of the evidence supporting the hypothesis
81 of mixing-triggered eruptions in the Aeolian archipelago, providing textural, chemical, and
82 rheological constraints.

83 Subsequently, we experimentally investigated the AD 1739 eruption at Vulcano Island
84 (Aeolian archipelago), chosen as a case study because it represents one of the most active and
85 dangerous volcanoes in the Aeolian archipelago, posing the highest risk in the region (De Astis et
86 al., 2013). By comparing the results of magma mixing experiments to natural samples, we provide
87 an estimate of the timescales occurring between the beginning of the mixing process in the shallow
88 magmatic reservoir and the subsequent eruption. Finally, we present an estimate of the averaged
89 magma ascent rates, and discuss the implications of the results for an improved definition of the
90 dynamics occurring in the Vulcano plumbing system prior to the eruption.

91

92 **2. Geological setting and petro-volcanological background**

93 *2.1. Geologic summary of the Aeolian archipelago and evidence of magma mixing*

94 The Aeolian Islands are a volcanic archipelago in the Southern Tyrrhenian Sea, north of the
95 coast of Sicily. The archipelago consists of seven islands: Lipari, Stromboli, Vulcano, Salina,
96 Alicudi, Filicudi, and Panarea (Fig. 1A). The subaerial volcanoes of this archipelago and the
97 associated volcanic seamounts form a ring-shaped structure emplaced on continental crust.

98 The volcanic products (aged from 430 ka to present; Peccerillo, 2005) range from calc-
99 alkaline, high-K calc-alkaline, to shoshonitic and alkaline (Keller, 1980; Peccerillo, 2005; Fig. 2).

100 The geochemical affinity of these rocks and the occurrence of deep seismicity in the
101 southern sector of the Tyrrhenian Sea connects the genesis of the Aeolian archipelago to the
102 subduction of the Ionian domain beneath the Calabrian arc (Keller, 1980; Peccerillo, 2005).

103 The mixing process, together with fractional crystallization, magma–fluid interactions, and
104 different degrees of host rock assimilation, has often been invoked as a recurring event in the
105 petrological evolution of magmatic feeding systems throughout the Aeolian arc (e.g., De Rosa et al.,
106 2003; Francalanci et al., 1989; Gioncada et al., 2005; De Astis et al., 2013). Many studies
107 addressing magma mixing have focused on detailed descriptions and analysis of morphological
108 features (De Rosa et al., 1996, 2002, 2003; Perugini et al., 2004, 2007; Ventura et al., 2006; Vetere
109 et al., 2015a) or physicochemical disequilibria in crystals (Landi et al., 2004; Perugini and Poli,
110 2012). Furthermore, particular emphasis has been given to the study of mafic enclaves dispersed in
111 more felsic magmas, as these occur in several outcrops throughout the Aeolian archipelago (e.g.,
112 Pollara depression and Porri lava flow at Salina; La Fossa cone at Vulcano; Monte Guardia and
113 Monte Pilato at Lipari; De Rosa et al., 2003; Perugini et al., 2004, 2007; Piochi et al., 2009; Davì et
114 al., 2009, 2010, Vetere et al., 2015a).

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116 *2.2. Vulcano Island and the AD 1739 and 1888–90 eruptions*

117 Vulcano is the southernmost island of the Aeolian archipelago and, along with Salina and
118 Lipari, it forms a NNW–SSE-striking volcanic belt deviating from the Aeolian ring-shaped
119 structure (Peccerillo, 2005; Fig. 1B).

120 The southern part of Vulcano Island is formed by the Primordial Vulcano and Piano
121 Caldera, while the northern sector hosts the Mastro Minico-Lentia complex and the Fossa Caldera
122 (De Astis et al., 2013). The volcanic activity of Vulcano started ca. 120 ka ago (Peccerillo, 2005;
123 Clocchiatti et al., 1994; De Astis et al., 2013; Fig. 1B) and the last eruptions occurred in AD 1739

124 and 1888–90. The Vulcanello Peninsula, along the northern part of the island, consists of
125 shoshonitic and latitic lava flows. Along with the volcanic products of the La Fossa cone, these
126 products represent the most alkaline compositions found in the Aeolian archipelago (Peccerillo,
127 2005; De Astis et al., 2013). In particular, the magmatic products from the AD 1739 and 1888–90
128 eruptions range from latitic to rhyolitic compositions (Peccerillo, 2005; Clocchiatti et al., 1994;
129 highlighted in red as probability density function in Fig. 2A).

130 Two different phases characterize the AD 1739 eruption. The early explosive phase was
131 essentially phreatomagmatic. It produced several low-energy pyroclastic density currents (PDCs)
132 and minor fallout episodes (De Astis et al., 2013). The deposits generated during this phase are
133 prevalently exposed along the flanks of La Fossa cone and are interlayered with banded pumiceous
134 trachy-rhyolitic bombs.

135 The subsequent effusive phase produced the Pietre Cotte lava flow (Fig. 3A). It represents
136 the latest lava flow event at Vulcano Island and shows evidence of magma mixing through the
137 presence of mafic enclaves in a rhyolitic host (Piochi et al., 2009). Detailed observations suggest
138 that the mafic enclaves constitute 5–7% of the entire outcrop and their presence led to the
139 interpretation of the occurrence of magma mixing during the eruptive event (Perugini et al., 2007).

140 The AD 1888–90 eruption represents the latest activity of La Fossa (Clocchiatti et al., 1994;
141 De Astis et al., 2013). The initial eruptive phases are recorded in the stratigraphic sequence by
142 rhyolitic phreatomagmatic breccias containing magmatic enclaves (latitic in composition). The
143 following phases were predominantly explosive and consisted of pyroclastic surges and intermittent
144 fallout products ranging from trachytic to latitic. The upper sequence is characterized by rhyolitic
145 tephra, similar in composition to the initial eruptive phase (Clocchiatti et al., 1994; De Astis et al.,
146 2013). The occurrence of magma mixing is highlighted by reaction rims in the different phases,
147 reversely zoned clinopyroxenes, banded pyroclastic products, and the presence of magmatic
148 enclaves (latitic in composition). Clocchiatti et al. (1994) fixed the maximum volume of the latitic
149 enclaves in the hosting rhyolite to ~10%.

150 The estimated temperature for mixing processes, which occurred in a shallow reservoir (~ 1
151 kbar; De Astis et al., 2013) for both the AD 1739 and 1888–90 events, is 1000–1100°C for the latite
152 and 955–1000°C for the rhyolite, in agreement with Vetere et al. (2015a) and Clocchiatti et al.
153 (1994), respectively. To achieve such conditions, the volumetric proportions of latitic magma
154 involved in the re-heating of the shallow magmatic reservoir were assumed to be significantly
155 higher than the amount shown by field observations as reported by Vetere et al. (2015a). This
156 observation also agrees with the experimental results reported by Laumonier et al. (2014b), stating
157 that onset of mixing in a felsic reservoir (e.g., rhyolitic or dacitic) requires a mass fraction of mafic
158 magma of approximately 60–65% (Laumonier et al., 2014b).

159

160 **3. Methodology**

161 *3.1. Experimental end-members and banded pumices*

162 The utilized end-members were sampled by the Petro-Volcanology Research Group (PVRG;
163 <http://pvrg.unipg.it>), University of Perugia. The least evolved was a latite sampled from the top part
164 (AD 1888–90 eruption) of the La Fossa cone, while the most evolved was a high-K rhyolitic
165 obsidian from the Pietre Cotte lava flow (AD 1739 eruption). The choice of the end-members was
166 determined by field observations and geochemical modeling of whole-rock analysis pointing to
167 these compositions as end-members for the mixing process occurring before the AD 1739 eruption
168 (Clocchiatti et al., 1994; Piochi et al., 2009; Vetere et al., 2015a).

169 The rhyolitic end-member (i.e. felsic; Fig. 3B) is characterized by a banded aspect with an
170 alternation of pumiceous and glassy bands with elongation parallel to the lava flow direction
171 (Vetere et al., 2015a).

172 The latitic end-member (i.e. mafic; Fig. 3C) appears as a dark gray rock showing a high
173 degree of vesiculation. It occurs as enclaves within the rhyolitic host. Petrographically, it shows a
174 porphyritic texture with euhedral to sub-euhedral crystals of clinopyroxene and plagioclase

175 immersed in a microcrystalline groundmass composed of clinopyroxene, plagioclase, opaque
176 minerals, and glass.

177 Fig. 4 shows the composition of the end-members plotted in a total-alkali vs. silica diagram
178 (TAS, Le Maitre, 1989) in which the literature data for Vulcano Island are also reported.

179 Banded pumices, utilized to compare the experimental results to natural samples, are from
180 the fallout events of the early explosive phase of the AD 1739 eruption (Fig. 3D). Macroscopically,
181 they show a high degree of vesiculation along with the presence of several filament-like structures.
182 Optical analysis showed a glassy matrix with a few occurrences of small phenocrysts (mode <2%),
183 mainly plagioclase, clinopyroxene, and minor olivine. The collected banded pumices did not
184 highlight significant inter-sample textural and compositional variations. Because of these
185 similarities, we selected one representative sample to be utilized in the present study. The sample
186 was prepared for a 100- μm thin section and polished using up to 1- μm diamond pastes for the
187 analysis via electron microprobe.

188

189 *3.2. Experimental apparatus and methods*

190 To provide mixing-to-eruption timescales, new magma mixing experiments were performed
191 using the Chaotic Magma Mixing Apparatus (COMMA, Morgavi et al., 2015) installed at the
192 Department of Physics and Geology, University of Perugia (Italy). The device was designed to
193 conduct experiments at high temperature with high-viscosity (up to 10^6 Pa s) natural and synthetic
194 magmas up to a temperature of 1800°C (Fig 5; Morgavi et al., 2015).

195 The apparatus produces chaotic flow dynamics in the mixing system using a stirring
196 protocol known as a Journal Bearing System (JBS; Swanson and Ottino, 1990). The ability of the
197 JBS to produce efficient mixing patterns, by fully controlling the mixing kinematics, makes it well
198 suited for the experimental investigation of magma mixing processes (De Campos et al., 2008,
199 2011; Morgavi et al., 2015; Rossi et al., 2017). The utilized mixing protocol produces textures
200 representative of several natural examples bearing evidence of magmatic interaction, among which

201 many cases have been documented in the Aeolian Islands (De Rosa et al., 2002; Perugini et al.,
202 2004; Perugini and Poli, 2012). Therefore, it allows one to reproduce the non-linear nature of the
203 mixing process (e.g., chaotic; Flinders and Clemens, 1996; De Rosa et al., 2002). In addition, the
204 aforementioned cited study cases support the robustness of the assumption that chaotic mixing
205 experiments can accurately reproduce pre-eruptive magma mixing events (De Campos et al., 2008,
206 2011; Perugini et al., 2015).

207 The system is composed of a Pt₈₀Rh₂₀ crucible (i.e. the outer cylinder) that contains the two
208 end-member compositions and an inner, off-centred spindle (i.e. the inner cylinder; Fig. 5). The
209 geometry of the system is determined by two parameters: (a) the ratio of the radii of the two
210 cylinders $r = R_{in}/R_{out} = 1/3$ and (b) the eccentricity ratio of the outer cylinder $\varepsilon = \delta/R_{out} =$
211 0.3, where δ is the distance between the centers of the two cylinders (Swanson and Ottino, 1990).
212 The mixing of magmas is triggered by the alternating rotations of the inner and the outer cylinder in
213 opposite directions because both streamlines and velocity are time-dependent (Galaktionov et al.,
214 2002; Swanson and Ottino, 1990). For the present work, the characteristic velocity $V = |V_{in}| +$
215 $|V_{out}|$ and characteristic length $L = R_{out} - R_{in}$ are equal to $2.8 \times 10^{-5} \text{ m s}^{-1}$ and 1.23×10^{-2}
216 m, respectively, with a strain rate of $\sim 10^{-3} \text{ s}^{-1}$, calculated in agreement with Galaktionov et al., 2002.

217 During the experiments, the relative volumes of the felsic (rhyolite) and mafic (latite) end-
218 members were 88% and 12%, respectively. Notably, the volumetric proportions used in the
219 experimental setup were relevant to those observed in the erupted products (Perugini et al., 2007;
220 Clocchiatti et al., 1994), although the volumetric balances in the magmatic reservoir were
221 reasonably different with much greater volume of the less-evolved end-member being intruded (see
222 also section 2.2).

223 The durations of the two experiments were 10.5 h (Experiment A) and 31.5 h (Experiment
224 B), corresponding to one and three repetitions of a protocol consisting of two rotations of the
225 crucible over 7 h and six rotations of the spindle over 3.5 h, respectively. The experiment started

226 with two homogeneous contrasting multicomponent silicate melts. With time, the initial elemental
227 concentrations in the end-members were expected to drift towards the hybrid of the system.

228 The experimental conditions were $P = 1 \text{ atm}$ and $T = 1200^\circ\text{C}$. The choice of 1200°C for the
229 experiments aimed at compensating for the absence of water during the experiments performed at
230 atmospheric pressure, resulting a rheological and chemical behaviour similar to the natural
231 occurrence ($\text{H}_2\text{O} \sim 1.5 \text{ wt.}\%$; $T \sim 1000^\circ\text{C}$; $P \sim 1 \text{ kbar}$; Clocchiatti et al., 1994). In particular, we
232 utilized a parametrization for the viscosity (η) and the chemical diffusivities based on the models of
233 Giordano et al. (2008) and González-García et al. (2017), respectively. In detail, Fig. 6 shows the
234 parametrization for the viscosity based on the model of Giordano et al. (2008). For both the latitic
235 and rhyolitic end-members, the value of η in the presence of $\sim 1.5 \text{ wt.}\% \text{ H}_2\text{O}$ at $T=1000^\circ\text{C}$ (i.e. the
236 estimated temperature for the AD 1739 eruption) agrees with the parametrization of η for dry melts
237 at the experimental temperature (i.e. 1200°C). For the chemical diffusivities, the presence of ~ 1.5
238 wt.% of water leads to an increase in the chemical diffusivities of $\sim 1.5 \text{ log units}$ compared to that of
239 dry conditions (González-García et al., 2017). Assuming an Arrhenian behaviour for the selected
240 chemical species (e.g., Zhang et al., 2010), an increase in the temperature of $\sim 200^\circ\text{C}$ well agrees
241 well with an increase in the chemical diffusivities of $\sim 1.5 \text{ log units}$.

242 The temperature inside the experimental sample was monitored using an S-type Pt/Pt-Rh
243 thermocouple. The error for the temperature was $\pm 3^\circ\text{C}$.

244 At the end of each experiment, the furnace was rapidly moved downward to allow the
245 experimental sample to be near the cooling heads positioned at the top and the bottom exits of the
246 furnace tube. To ensure the achievement of a glassy sample, we calculated the NBO/T value in
247 agreement with Mysen and Richet (2005) for the latitic composition (i.e. the experimental
248 composition in which the development of crystals preferentially occurs in the studied system). The
249 resulting value was 0.27. Starting from the obtained NBO/T value, we calculated the critical cooling
250 rate (R_c) as ca. $4000 \text{ }^\circ\text{C/h}$, according to Vetere et al., (2015b). During the quenching, the
251 temperature of the samples decreased from 1200°C to 665°C (T_g for our composition calculated

252 after Giordano et al., 2008) in ca. 3 min, corresponding to a quench rate of $\sim 10^4$ °C/h, significantly
253 higher than the obtained Rc. This rate completely suppressed nucleation and crystallization
254 processes within the experimental samples. The sample was then cored out from the outer cylinder,
255 set in epoxy resin, and left to harden for 24 h. Ultimately, the experimental product was cut into
256 slices ca. 4 mm in thickness and polished using up to 1- μ m diamond pastes for analysis using an
257 electron microprobe.

258

259 *3.3. Analytical methods*

260 Major element concentrations and back-scattered electron (BSE) images were acquired
261 using a Cameca SX100 electron microprobe (EMP) at the Institut für Mineralogie, Leibniz
262 Universität Hannover (Germany). The operating conditions were characterized by an acceleration
263 voltage of 15 kV, a beam current of 4 nA, and a beam diameter of 10 μ m to minimize the alkali loss
264 in the glass analysis. Raw data were converted to concentration using the software “Peak Sight” and
265 “PAP” matrix (Pouchou and Pichoir, 1991). The reference materials utilized for the calibration were
266 wollastonite for Ca and Si, periclase for Mg, hematite for Fe, corundum for Al, rutile for Ti,
267 orthoclase for K, and albite for Na. The counting time was 10 s for each element.

268 Accuracy and precision were determined for the VG-568 (rhyolite) and VG-2 (basalt)
269 glasses (Jarosewich et al., 1980), utilized as secondary reference materials. Absolute analytical
270 errors, measured for international reference materials, were of the order of 4% for all analyzed
271 elements. During the experiments, a total of five profiles (two for experiment A and three for
272 experiment B) were obtained with a spacing of 15 μ m between the analysis points, for a total of 877
273 analytical determinations. The same analytical conditions were used for the three transects
274 performed on the banded pumice, for a total of 92 analytical determinations.

275

276 *3.4. Mixing to Eruption Timescales estimate*

277 Mixing-to-eruption timescale estimates were developed using the method of concentration
 278 variance decay (CVD; Perugini et al., 2015). It was utilized to capture the time elapsing between the
 279 beginning of the mixing process (i.e. the injection of new mafic magma in the reservoir) and the
 280 eruption. It consisted of experimentally evaluating the rate at which the concentration variance of
 281 the different chemical elements decayed with time during the mixing process (e.g., Perugini et al.,
 282 2015). The successive step consisted of the derivation of empirical relations, typically exponential,
 283 linking the variance decay with time. Finally, comparing the concentration variances of the natural
 284 sample, namely the banded pumice generated by the fallout events of the early explosive phase of
 285 the AD 1739 eruption (Fig. 3D), to the parametrization derived from the experiments, we provided
 286 an estimate of the mixing-to-eruption timescales.

287

288 3.4.1. Concentration variance and concentration variance decay

289 In the fluid dynamics literature, concentration variance is commonly used to evaluate the
 290 degree of homogenization in fluid mixtures (e.g., Rothstein et al., 1999). For a given chemical
 291 element, the concentration variance is defined as follows:

$$292 \sigma^2(C^i) = \frac{\sum_{n=1}^N (C_n^i - \mu^i)^2}{N} \quad (1)$$

293 where N is the number of the analytical determinations, C_n^i is the concentration of a given chemical
 294 element i in the analytical determination n , and μ^i is the average composition for the element i . The
 295 concentration variance is strongly dependent on the mixing time: when the mixing time increases
 296 and the system moves towards homogeneity, the concentration variance progressively decreases.
 297 Moreover, $\sigma^2(C^i)$ depends on the concentrations of each element i .

298 In this work, considering the concentration of chemical elements in the experimental sample
 299 reported in Table 1 and following the method of Perugini et al. (2015), we normalized the variance
 300 values to the initial variance utilizing the following equation:

$$301 \sigma_n^2(C^i) = \frac{\sigma^2(C^i)_t}{\sigma^2(C^i)_{t=0}} \quad (2)$$

302 where $\sigma^2(C^i)_t$ and $\sigma^2(C^i)_{t=0}$ are, respectively, the variance in the chemical element (C^i) at time t
303 and $t = 0$ (i.e. before the beginning of the experiment or mixing event in nature).

304 In detail, a concentration variance (σ_n^2) was calculated for all the investigated chemical
305 elements in each analyzed transect. The operation was repeated for the two-mixing times (i.e. 10.5 h
306 and 31.5 h). The obtained values were subsequently averaged to obtain a representative estimation
307 of the homogenization degree in the studied system.

308 In addition, using the same procedure, we calculated the σ_n^2 for the natural sample (the
309 banded pumice sampled from the fallout deposits of the AD 1739 eruption). Hereafter, we always
310 refer to the concentration variance considering Eq. (2).

311 The concept of concentration variance decay (CVD) was used in Perugini et al. (2015) and
312 Rossi et al. (2017) in the assessment of the elemental mobility and chemical homogenization of
313 silicate melt systems. Particularly, for all the analyzed chemical elements, the CVD is modeled by
314 an exponential law as follows:

$$315 \sigma_n^2(C^i) = C_0 \exp(-Rt) + C_1 \quad (3)$$

316 where C_0 and C_1 represent the value of the concentration variance [$\sigma_n^2(C^i)$] at $t = 0$ and $t = \infty$,
317 respectively, and t is the mixing time. The fitting parameter R is termed the “relaxation of
318 concentration variance” (RCV) and quantifies the rate at which the concentration variance decays
319 with increasing mixing time and, in addition, the mobility of the chemical elements in the system.
320 Regarding the mobility of the chemical elements, the main controlling parameters in a melt are the
321 temperature, chemical composition, volatile content, oxygen fugacity, and pressure (e.g., Zhang et
322 al., 2010). In addition the R parameter accounts, in a single variable, for a range of processes
323 affecting elemental mobility including the (a) partitioning of the chemical elements into different
324 melts (Watson, 1976), (b) influence of advection on apparent diffusive fluxes (Perugini et al.,
325 2006), (c) dependence of diffusivities on multicomponent composition (e.g., Zhang et al., 2010),

326 and (d) potential development of “uphill” or “downhill” diffusion processes (Watson and Jurewicz,
327 1984).

328

329 **4. Results and Discussion**

330 *4.1. Evidence linking magma mixing to the eruptive events of the Aeolian archipelago*

331 The main significant points resulting from the review of current literature highlighted that i)
332 magma mixing has been widely documented for most of the eruptions occurring in the Aeolian
333 Islands, ii) many studies have linked the mixing process to an eruption, iii) many works have noted
334 the rapid remobilization of a shallow reservoir prior to an eruption (Table 2 and references therein).
335 In the following, we report the main findings resulting from the investigation of literature data.

336 At Salina Island, the interaction between a dacitic and a basaltic-andesite magma has been
337 shown by the presence of magmatic enclaves in the Porri lava flow (ca. 43 ka; De Rosa et al., 1996;
338 Ventura et al., 2006) and in products from the Upper Pollara eruption (ca. 13 ka; Perugini et al.,
339 2004). In detail, Perugini et al. (2004) underlined that the analysis of concentration patterns of
340 heterogeneous juvenile products allows for the estimation of initial magma concentrations, and also
341 providing constraints for pre- and syn-eruptive dynamics. In addition, Perugini et al. (2004)
342 interpreted the whole eruptive sequence as the result of the emptying of the whole magmatic
343 reservoir, highlighting that the flow regime during the interaction between the two end-members
344 was heterogeneous, from sub-turbulent to completely turbulent, just prior to and during the
345 eruption.

346 Regarding the Porri lava flow, De Rosa et al. (1996) and Ventura et al. (2006) reported its
347 structural and geochemical features with evidence of the interaction between two different magmas.
348 They utilized the mixing structures (e.g., enclaves and bandings) to calculate the shear strain of the
349 lava flow and to evaluate the efficiency of the process (De Rosa et al., 1996). Particularly, Ventura
350 et al. (2006) made inferences on the degree of interaction between the different magmas on the
351 basis of the spatial distribution of the enclaves. According to this hypothesis, host magmas

352 characterized by a homogeneous spatial distribution of enclaves are indicative of efficient advection
353 processes that may favor magma mixing. In contrast, host magmas characterized by an
354 inhomogeneous distribution of enclaves suggest a low dynamical interaction between the two end-
355 members, and therefore inefficient mixing (Ventura et al., 2006).

356 At Lipari, the juvenile material recognized in the products of the Monte Guardia rhyolitic
357 eruption (ca. 22 ka) was investigated by petrographic, mineralogical and melt inclusion-based
358 studies (De Rosa et al., 2003). Results highlighted that different mixing textures were produced in a
359 single mixing event before the first stage of the eruption, followed by a second phase characterized
360 by a larger volume of magma of rhyolitic composition (De Rosa et al., 2003). Moreover, chemical
361 analysis of zoned clinopyroxenes and plagioclases found in these rocks (Gioncada et al., 2005)
362 support the idea that the chemical zonation in the minerals preceded the eruptive event, with a
363 residence time before mixing on the order of decades to hundreds of years (Gioncada et al., 2005).

364 Additionally, Davì et al. (2009, 2010) focused on latitic enclaves found in the rhyolitic lava
365 flow of Rocche Rosse, produced by the AD 1230 Monte Pilato eruption. The estimated viscosities
366 were then applied to rheological modeling of mixing processes between the latitic and rhyolitic
367 magmas at conditions relevant for the Rocche Rosse eruption. Results suggested rapid ascent
368 events, occurring just after the beginning of the mixing process (Davì et al. 2009, 2010).

369 At Stromboli, magma mixing has been widely investigated by several authors (Francalanci
370 et al., 1989, 2012; Petrone et al., 2006, 2018; Armienti et al., 2007; Landi et al., 2004).
371 Mineralogical and chemical studies of lavas and pyroclastic products produced during the 1985–
372 2000 and 2002–2003 eruptions highlighted mixing-related chemical zoning in plagioclase
373 phenocrysts (Landi et al., 2004, 2009). In detail, evidence suggested recharge events in the
374 magmatic system by hotter magma batches. These mixing events, often developing between
375 magmas of similar mafic composition, efficiently occurred in the Stromboli feeding system just
376 before the eruptions during the period 1985–2000 as highlighted by the disequilibrium textures in

377 the plagioclase phenocrysts (showing an alternation of labradoritic and bytownitic layers; Landi et
378 al., 2004, 2009).

379 Textural evidence of magma mixing was also reported by Armienti et al. (2007), defining
380 significant chemical zoning and resorption textures in plagioclase, olivine, and pyroxene crystals. In
381 particular, a crystal size distribution analysis suggested a stationary crystallizing system being
382 continuously re-injected by crystal-free basaltic melt (Armienti et al., 2007). In addition, the large
383 radiogenic isotope (Sr, Nd, Pb, U, and Th) variations detected in the plagioclase, olivine, and
384 clinopyroxene (Francalanci et al., 1999, 2012; Landi et al., 2009) confirmed the occurrence of
385 mixing processes between the crystal-rich and crystal-poor magmas in a shallow magmatic
386 reservoir (Francalanci et al., 2012). Moreover, Petrone et al. (2006) highlighted oscillatory patterns
387 and sharp compositional changes in clinopyroxene and plagioclase phases, suggesting complex
388 dynamics in the magmatic reservoir. Petrone et al. (2006) also suggested the injection of mafic
389 magma batches into the magmatic reservoir, highlighting that magma mixing played a significant
390 role in the development of compositional changes and temperature variations (Petrone et al., 2006).
391 Recently, Petrone et al. (2018) estimated the residence times of mafic magmas in the volcanic
392 feeding systems of the Post-Pizzo pyroclastic sequence (ca. 1.7–1.5 ka) and Early Paroxysms of the
393 present eruptive sequence using Fe–Mg diffusion in clinopyroxenes. They provided timescales of
394 the order of 1 to approximately 50 years between the arrival of the mafic magma in the shallow
395 system and the eruption (Petrone et al., 2018). In addition, modeling provided by Petrone et al.
396 (2018) highlighted mixing-to-eruption timescales from days to months for selected clinopyroxenes,
397 further suggesting that rapid mixing events and short storage timescales are key parameters in the
398 pre-eruptive dynamics of the system. Many authors noted the evidence that the present-day activity
399 of Stromboli is controlled by a steady-state feeding system in which injection, mixing, degassing,
400 and crystallization at a shallow level continuously occur (Francalanci et al., 1999, 2012; Landi et
401 al., 2009; Petrone et al., 2006, 2018). Moreover, Landi et al. (2009) suggested that the change
402 between strombolian and effusive activity is mainly related to periods of relatively more or less

403 vigorous injection into the shallow magmatic system (Landi et al., 2009). As a summary, Table 2
404 reports a synthesis of the evidence linking the mixing process to the eruptive events of the whole
405 Aeolian archipelago.

406

407 *4.2. Evidence linking magma mixing to the AD 1739 and 1888–90 eruptions of Vulcano Island*

408 The process of magma mixing in the AD 1739 eruption of Vulcano Island has been widely
409 investigated (Perugini et al., 2007; Piochi et al., 2009; Vetere et al., 2015a), focusing on the
410 morphological, geochemical thermo-dynamical, and rheological characterization of the erupted
411 magmas.

412 All the studies noted the mixing process as having a primary role in the pre-eruptive
413 evolution of the volcanic plumbing system, and also suggesting a direct link between the occurrence
414 of magma mixing and the subsequent eruption. In detail, Perugini et al. (2007) quantitatively
415 investigated the morphology and size distributions of the enclaves in the Pietre Cotte lava flow.
416 Perugini et al. (2007) showed that the enclaves resulted from the fragmentation of viscous fingering
417 structures following the injection of mafic magma into a felsic reservoir just prior to the eruption. In
418 addition, geochemical investigations performed by Piochi et al. (2009) provided evidence that the
419 latitic enclaves and the rhyolitic host magma were subject to similar thermo-barometric conditions,
420 further supporting the idea of the coexistence of compositionally variable magmas in the shallow
421 magmatic reservoir beneath the La Fossa cone before the AD 1739 eruption.

422 Vetere et al. (2015a), using thermo-dynamical and rheological constraints, suggested that the
423 most plausible scenario for the AD 1739 eruption was a shallow reservoir in which a latitic magma
424 intruded a rhyolitic magma, finally leading to eruptive events. Part of the volume of the latitic
425 magma remained in the reservoir after the conclusion of the AD 1739 eruptive event, to be erupted
426 during the last cycle of AD 1888–90.

427 In the case of the AD 1888–90 eruptive cycle, Clocchiatti et al. (1994), utilizing whole-rock
428 analysis and isotopic investigations, highlighted that the erupted pyroclastic products were derived

429 from mixing between a rhyolitic end-member and a latite of similar composition to the enclaves
430 detected in the Pietre Cotte lava flow. In addition, Clocchiatti et al. (1994) highlighted a direct link
431 between the mixing process, occurring in the shallow reservoir, and the subsequent eruption. In
432 detail, Clocchiatti et al. (1994) suggested the unlocking of the rhyolite by re-heating resulting from
433 the injection of the hotter latite.

434

435 *4.3 Timescale estimation for the AD 1739 eruption*

436 The presented magma mixing experiments aimed to define the mixing-to-eruption
437 timescales, i.e. the time elapsed between the beginning of the mixing in the magmatic reservoir and
438 the subsequent eruption, for the AD 1739 eruption at Vulcano island.

439 The morphological results of the experiments are shown in Fig. 7, displaying BSE images of
440 selected portions of the experimental products in which the presence of stretched and folded
441 alternating filaments of the two end-members are clearly observable. The density of the filament
442 populations as well as their thicknesses strongly vary throughout the whole experimental product as
443 a direct consequence of the complex dynamics acting during the mixing process (i.e. the JBS
444 protocol; Swanson and Ottino, 1990).

445 The analysis performed using EMP on the experimental and natural samples is shown in
446 Table SM1. Fig. 8 shows the compositional variation of all the analyzed elements of the
447 experimental products (i.e. Al_2O_3 , CaO , MgO , K_2O , FeO , Na_2O , and TiO_2) vs. SiO_2 . End-member
448 compositions are also reported, as a reference.

449 Fig. 9 shows the evolution of the normalized concentration variance, i.e. $\sigma_n^2(C^i)$, in the
450 experiments as a function of time. In addition, Fig. 9 shows the results of the exponential fitting,
451 linking the evolution of the normalized variance and time (please refer to section 3.4.1. for further
452 details). Results highlight that the investigated elements (i.e. Si, Mg, Ca, Ti, Fe, Na, Al, and K)
453 show rates of decay, measured by the R parameter, ranging between 0.122 and 0.185 (Fig. 9).

454 Furthermore, Fig. 10 shows the variability in all the analyzed elements (i.e. Si, Mg, Ca, Ti,
455 Fe, Na, Al, and K) along transects made for the analyzed banded pumice used in this work as
456 natural sample, representative of the explosive phase of the AD 1739 eruption. The banded pumice
457 highlights compositions ranging between trachyte (i.e. $\text{SiO}_2 \sim 68$ wt.%) and rhyolite (i.e. $\text{SiO}_2 \sim 73$
458 wt.%; Fig. 10). These compositions agree well with those produced by the mixing process between
459 the latitic (i.e. $\text{SiO}_2 \sim 58$ wt.%) and rhyolitic (i.e. $\text{SiO}_2 \sim 74$ wt.%) end-members utilized during the
460 experiments, in the proportions reported in the present study (i.e. 12–88%).

461 As for the experimental products, the normalized variance was also estimated for the natural
462 sample, including all the analyzed elements (i.e. Si, Al, K, Na, Ca, Mg, Fe, and Ti). Mixing-to-
463 eruption timescales were then estimated using Eq. (3). The obtained mixing-to-eruption timescales
464 show a variability ranging from ca. 10 to 39 h (i.e. the minimum and the maximum value,
465 respectively), with an average of 29 ± 9 (s.d.) h (please refer to Table SM2 for further details). The
466 obtained timescales, of the order of days, are very short if compared to the few other estimates for
467 volcanic systems of the Aeolian archipelago ranging from years to decades (e.g., Gioncada et al.,
468 2005 and Petrone et al., 2018). However, they agree with the lower bound of the estimates reported
469 by Petrone et al. (2018), of the order of days to months, for clinopyroxene portions capturing the
470 final stages (i.e. final injections) of the mixing process at Stromboli.

471 In addition, the recovered mixing-to-eruption timescale estimates can be used to infer the
472 average magma ascent velocity in the case of the AD 1739 eruption of Vulcano Island. To achieve
473 such estimates, the definition of the depth for the final magma storage before the eruption was
474 essential. On the basis of petrological, geochemical, isotopic, and geobarometric data, De Astis et
475 al. (2013) showed that, in the case of the La Fossa cone, the most probable final storage region was
476 at ca. 3–5 km depth.

477 Assuming 3 and 5 km as the upper and lower boundary for the magmatic storage depths of
478 the AD 1739 eruption and combining them with the mixing-to-eruption timescales (i.e. 29 ± 9 h)
479 reported in the present manuscript, we estimated averaged magma ascent rates ranging from 3×10^{-2}

480 to $5 \times 10^{-2} \text{ m s}^{-1}$. Notably, the obtained ascent rate values are row estimates, and they do not imply
481 any inference regarding modulations of the velocity profile from the storage region to the surface
482 because of potential changes in the physical properties of magma during the ascent or other factors.

483 Finally, Fig. 11 shows the inferred temporal evolution of the shallow magmatic system
484 beneath the La Fossa cone before the AD 1739 eruption. In detail, at the initial time $t = 0$, a
485 rhyolitic reservoir was at ca. 3–5 km. The ascent of the latitic magma from a deeper reservoir (20–
486 25 km; De Astis et al., 2013) produced a re-heating of the rhyolite, triggering the unlocking of the
487 system and the mixing process, lasting for a maximum of ca. 39 h, and finally leading to the AD
488 1739 eruptive event.

489

490 **5. Concluding Remarks**

491 In this work, we extensively reviewed evidence supporting the hypothesis of mixing-
492 triggered eruptions for the Aeolian archipelago, providing textural, chemical, and rheological
493 constraints. Results highlighted a strict relation between the occurrence of magma mixing at
494 shallow crustal levels and the eruptions. In addition, the detailed review of literature data allowed us
495 to improve our understanding regarding pre-eruptive dynamics occurring in the feeding systems of
496 active volcanoes of the Aeolian archipelago.

497 Finally, the performed experiments provided constraints on mixing-to-eruption timescales
498 for the AD 1739 eruption that occurred at Vulcano Island, utilized as a case study. Our results
499 suggest an average mixing-to-eruption time (i.e. the time between the onset of mixing in the
500 reservoir and the eruption) of 29 ± 9 h and magma ascent rates ranging between 3×10^{-2} and 5×10^{-2}
501 m s^{-1} .

502 These new findings may have important implications for volcano monitoring and civil
503 protection purposes. For example, short mixing-to-eruption timescales and high ascent velocities
504 (e.g., Petrelli et al., 2018) for magma rising towards the Earth's surface may imply short warning
505 times for volcanic crises, on the order of days to hours.

506

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513

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721

722 **Figure Captions**

723 Figure 1: A) Schematic map of the Aeolian Islands and B) schematic geological map of Vulcano
724 Island (modified from Peccerillo et al., 2006) showing the main volcanic units and the location
725 of the Pietre Cotte lava flow generated during the effusive phase of the AD 1739 eruption.

726

727 Figure 2: A) Total alkali vs. silica diagram (TAS, Le Maitre, 1989) and B) SiO₂ vs. K₂O diagram
728 (Peccerillo and Taylor, 1976) showing the petrology of rocks of the Aeolian Islands. Literature
729 data are taken from Georoc ([http:// georoc.mpch-mainz.gwdg.de](http://georoc.mpch-mainz.gwdg.de)).

730

731 Figure 3: A) View of the Pietre Cotte lava flow; B) and C) end-members used in the mixing
732 experiments (B: rhyolite and C: latite); D) Example of banded pumice produced by the fallout
733 events of the early phase of the AD 1739 eruption used as natural sample in the present work.

734

735 Figure 4: Compositions of latitic (blue star) and rhyolitic (red star) end-members used in this work
736 plotted in a total-alkali vs. silica diagram (Le Maitre, 1989). Literature data for Vulcano Island

737 (taken from Georoc, [http:// georoc.mpch-mainz.gwdg.de](http://georoc.mpch-mainz.gwdg.de)) are also reported.

738

739 Figure 5: Schematic representation (not to scale) of the COMMA (reproduced and slightly modified
740 from Morgavi et al., 2015). A) Complete experimental setup showing the high-temperature
741 furnace, the upper and lower motors for the rotation of the spindle and the crucible hosting the
742 end-member melts, and the positioning of the experimental crucible; B) enlargement of the
743 experimental crucible filled with the end-member compositions; C) section of the crucible
744 perpendicular to its vertical axis showing the relative position of the end-members and the
745 spindle. Geometrical parameters of the used experimental protocol are also shown (see text and
746 Morgavi et al. (2015) for further details).

747

748 Fig. 6: Parametrization for the melt viscosity (η), based on the model of Giordano et al. (2008),
749 elaborated to compensate for the absence of water (1.5 wt.% in natural samples, Clocchiatti et
750 al., 1994) during the mixing experiments at atmospheric pressure. The gray area represents the
751 estimated η values under the physical conditions of the shallow reservoir before the AD 1739
752 eruption ($\sim 1000^\circ\text{C}$, ~ 1 kbar) (3.3 ± 0.2 Pa s for latite, and 5.1 ± 0.2 Pa s for rhyolite; Clocchiatti
753 et al., 1994). The blue line represents the calculated η values as a function of temperature. The
754 blue line intersects the gray area at $\sim 1200^\circ\text{C}$, the experimental temperature.

755

756 Figure 7: BSE images of selected areas of the experimental samples showing the intricate
757 morphologies produced during the mixing experiments. The location of the selected analyzed
758 transects is also reported. Please refer to Morgavi et al. (2013, 2015) for a detailed description
759 of the morphologies produced by JBS.

760

761 Figure 8: Inter-elemental binary diagrams for the analyzed chemical elements (CaO, MgO, K₂O,
762 Na₂O, FeO, Al₂O₃, and TiO₂) vs. SiO₂ showing the compositional variability resulting from

763 chaotic mixing processes during experiment A (10.5 h) and B (31.5 h). End-member
764 compositions are also reported, as a reference.

765

766 Figure 9: Variation in concentration variance as a function of mixing time for the analyzed chemical
767 elements fitted using Eq. (3) (see text). The value of R (i.e., the rate at which concentration
768 variance decays with time) is reported for each analyzed element.

769

770 Figure 10: Compositional variation of the analyzed chemical elements (SiO_2 , Al_2O_3 , CaO , MgO ,
771 Na_2O , K_2O , FeO , and TiO_2) along the transects analyzed for the banded pumice. Estimated
772 mixing-to-eruption timescales are reported for each analyzed element. The location of the
773 analyzed transects is also reported.

774

775 Figure 11: Schematic evolution (not to scale) of the shallow magmatic system beneath the La Fossa
776 cone before the AD 1739 eruption. A): The shallow magmatic reservoir is at 3–5 km depth
777 below the La Fossa cone (De Astis et al., 2013). Deeper batches of latitic magma start to rise
778 towards the felsic reservoir. B): At $t = 0$, the injection of a latitic magma into the rhyolitic
779 reservoir leads to the unlocking of the system and the chemo-physical interaction of the two
780 compositionally different magmas. C) After $t \geq 29$ h, the mixed magma reaches the surface
781 triggering the eruption.

782

783 **Table Captions**

784 Table 1: Concentrations (in weight %) and relative standard deviations of major elements in the
785 felsic and mafic end-member glasses used during the mixing experiments. The value N
786 represents the total number of analytical points averaged to obtain the reported concentrations.

787 The density (Ochs and Lange, 1999) and viscosity (Giordano et al., 2008) of both end-

788 members are also reported.

789

790 Table 2: Summary of the most relevant features (discussed in this work) detected in the rocks of the
791 Aeolian archipelago linking the mixing process to eruptive events.

792

793 **Supplementary material**

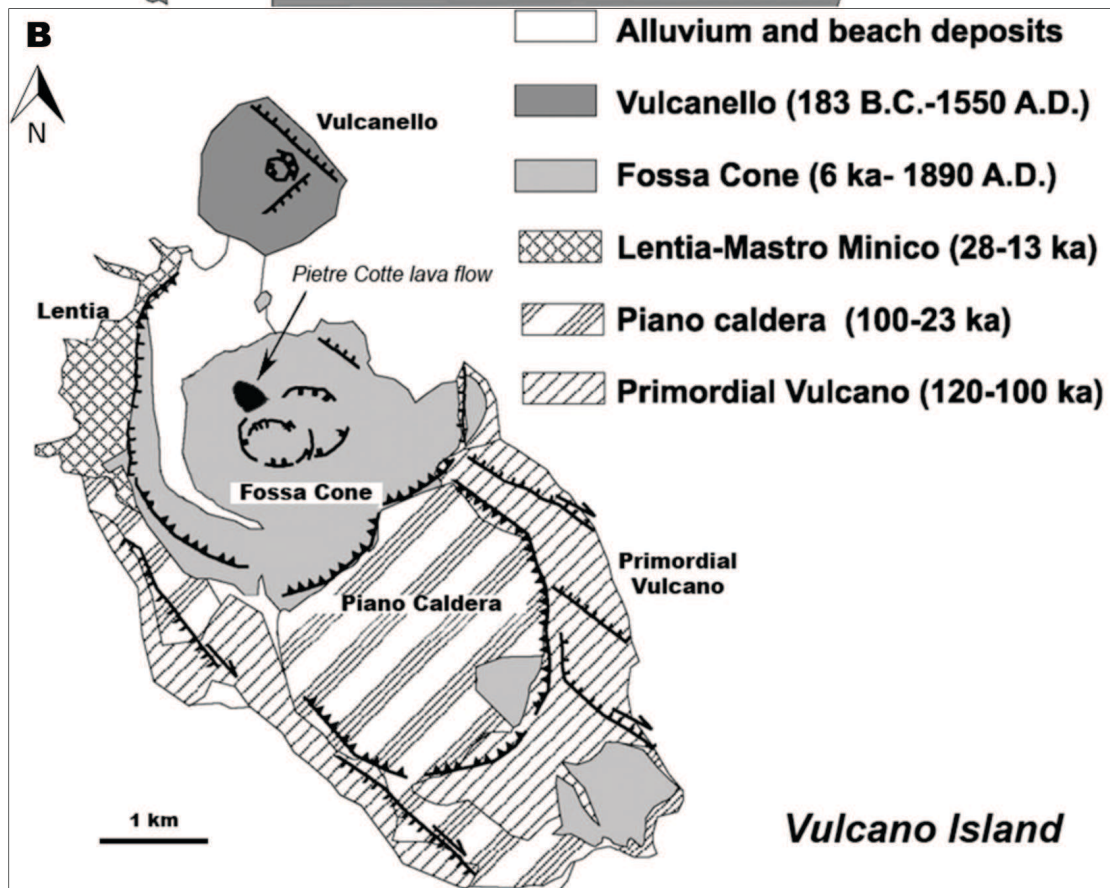
794 Table SM1: Complete analytical dataset containing all the analytical determinations performed on
795 the experimental (877 analysis) and natural (92 analysis) samples. A total of five and three
796 transects were analyzed for the experimental and the natural samples, respectively. The values
797 of the calculated concentration variance (σ^2) for each chemical element in each transect are
798 also reported.

799

800 Table SM2: Coefficients (C_1 , C_0 and R) used to fit the decay of the concentration variance for the
801 experimental samples and used to estimate the mixing-to-eruption timescales for the natural
802 sample.

803

Figure 1



Figure

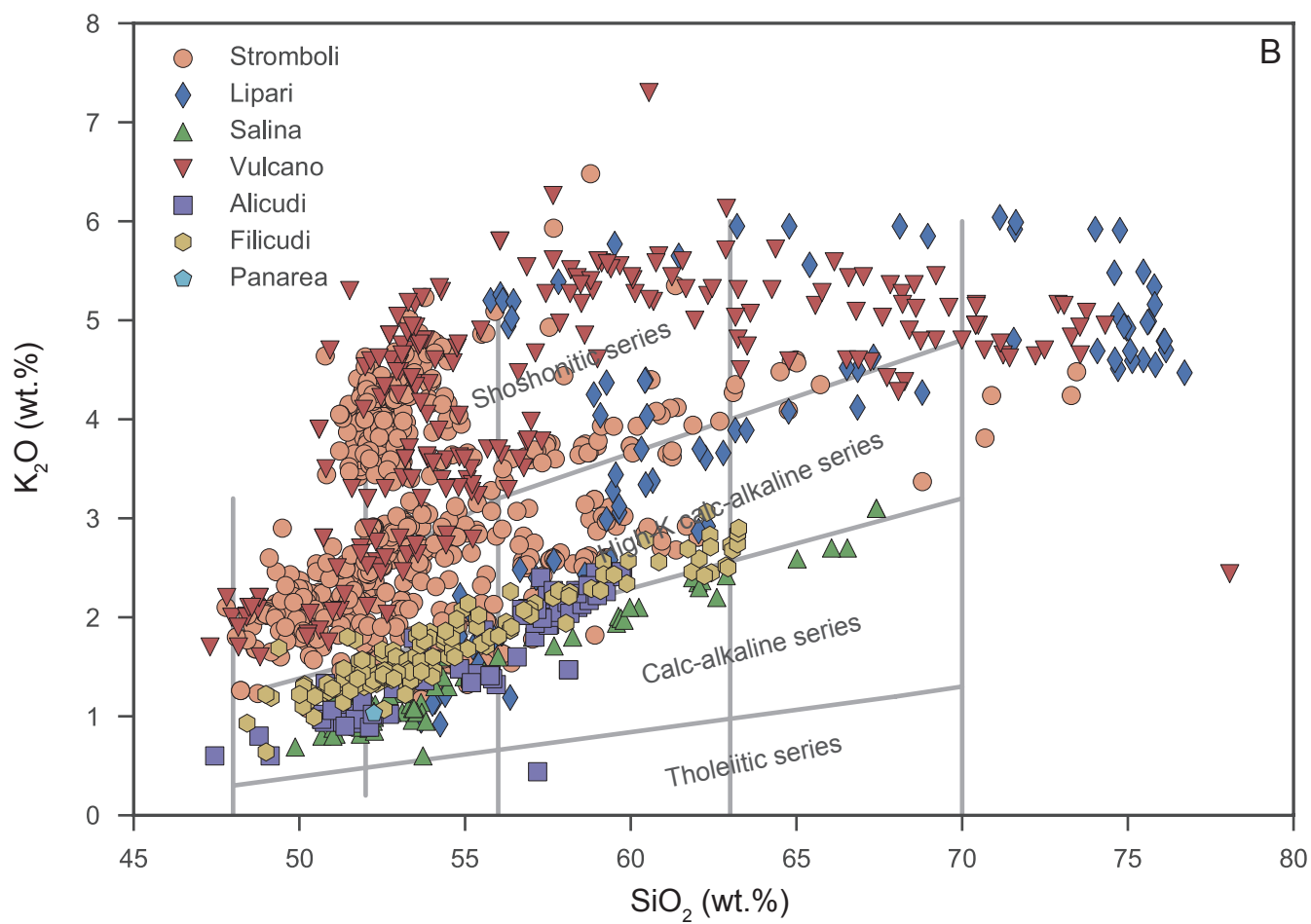
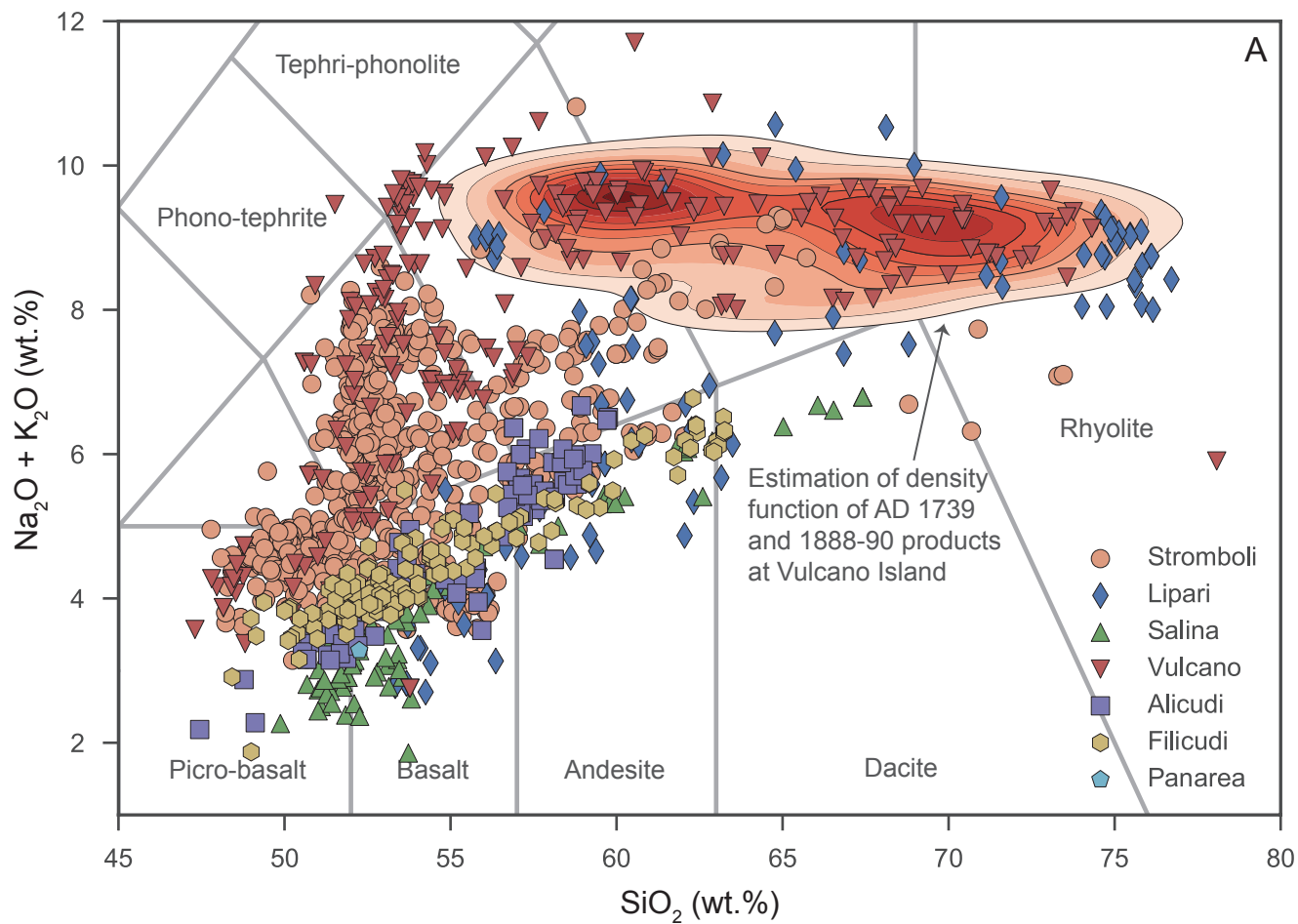


Figure 2

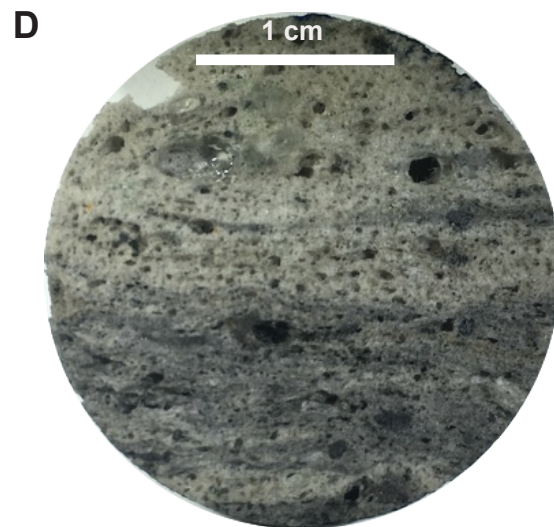


Figure 3

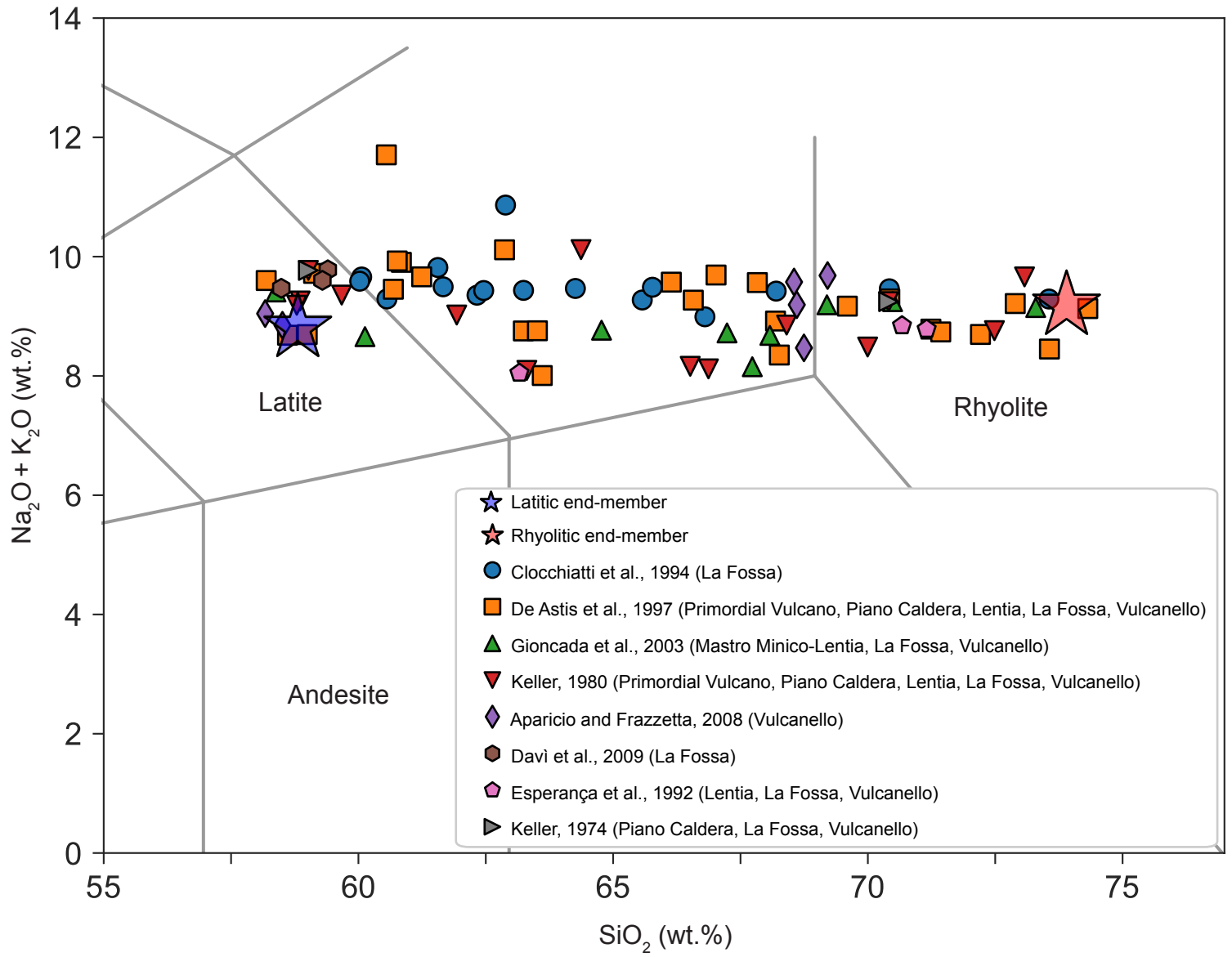


Figure 4

Figure

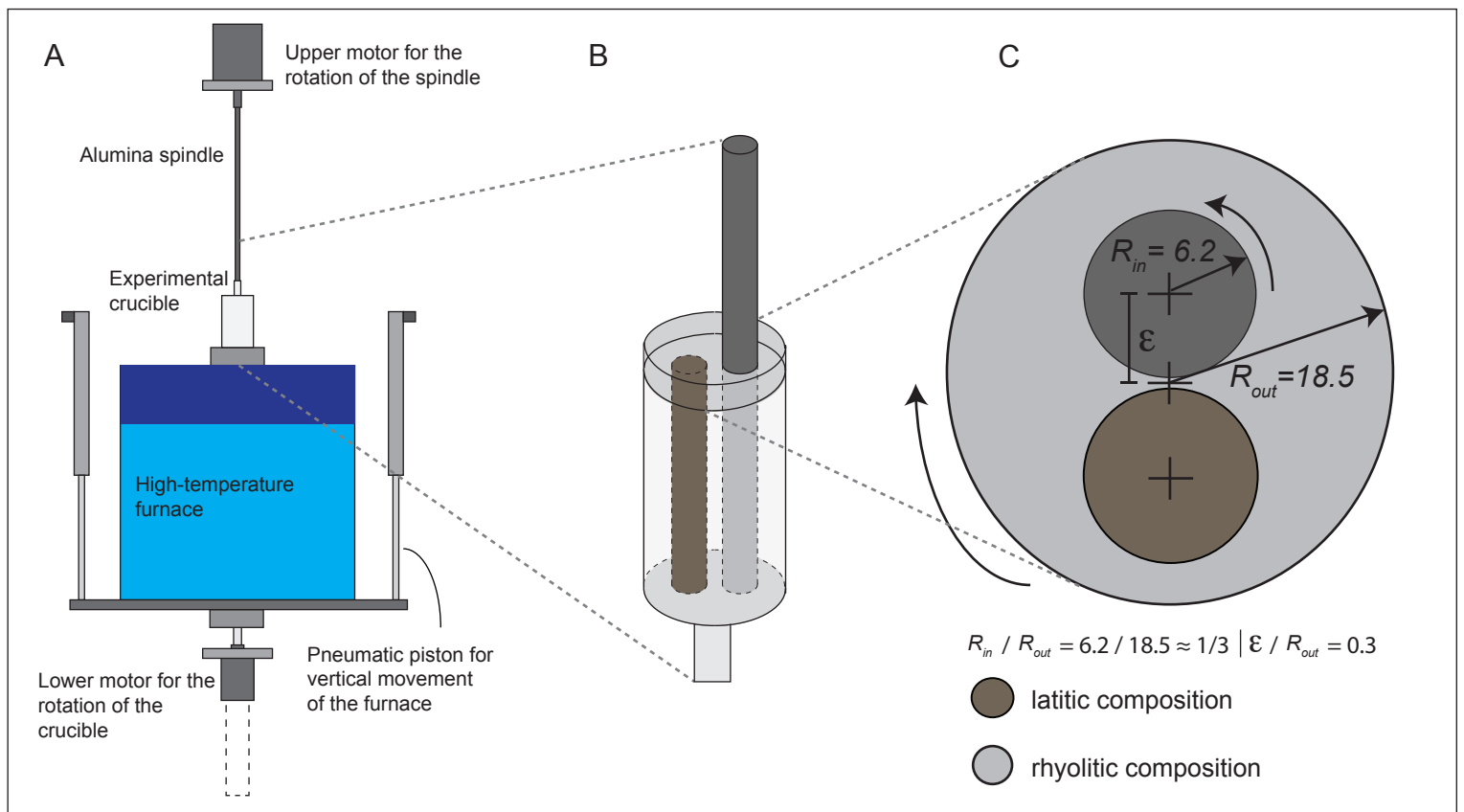
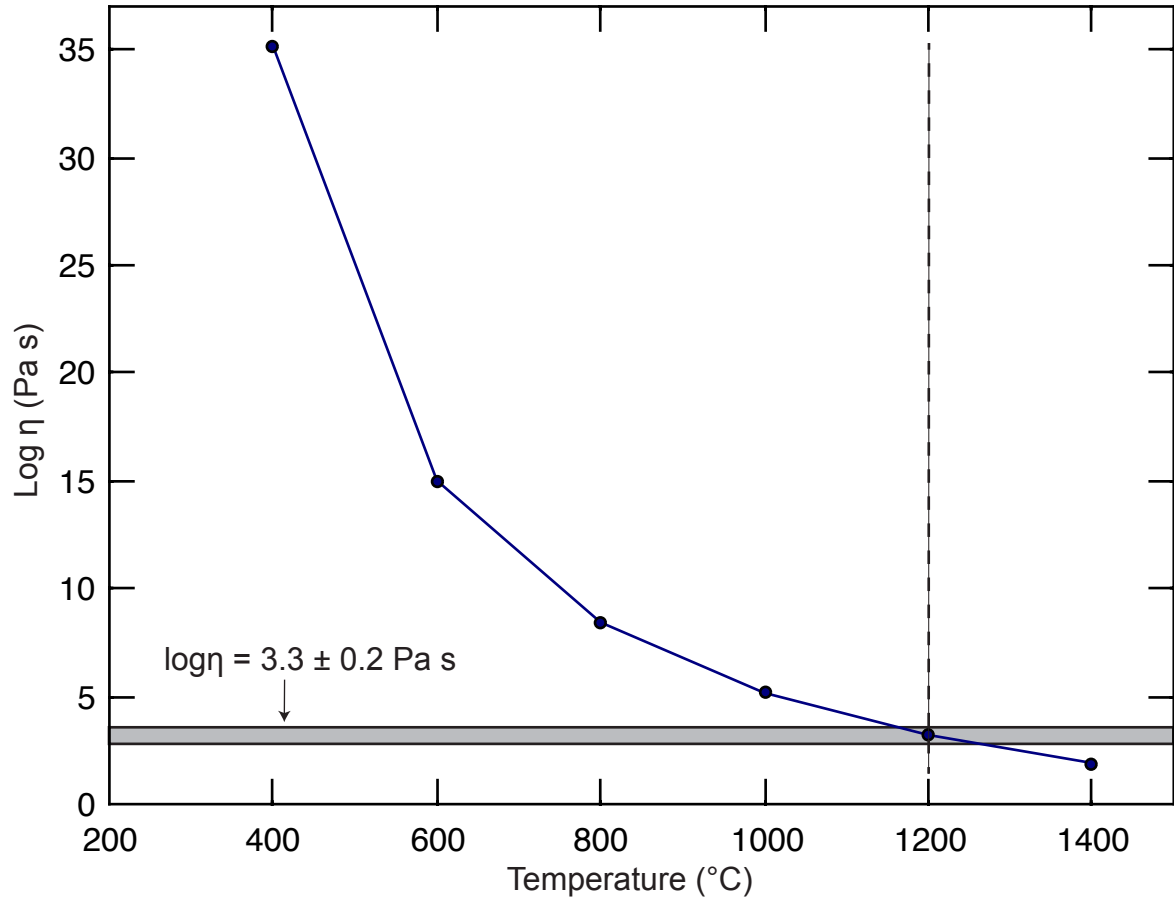


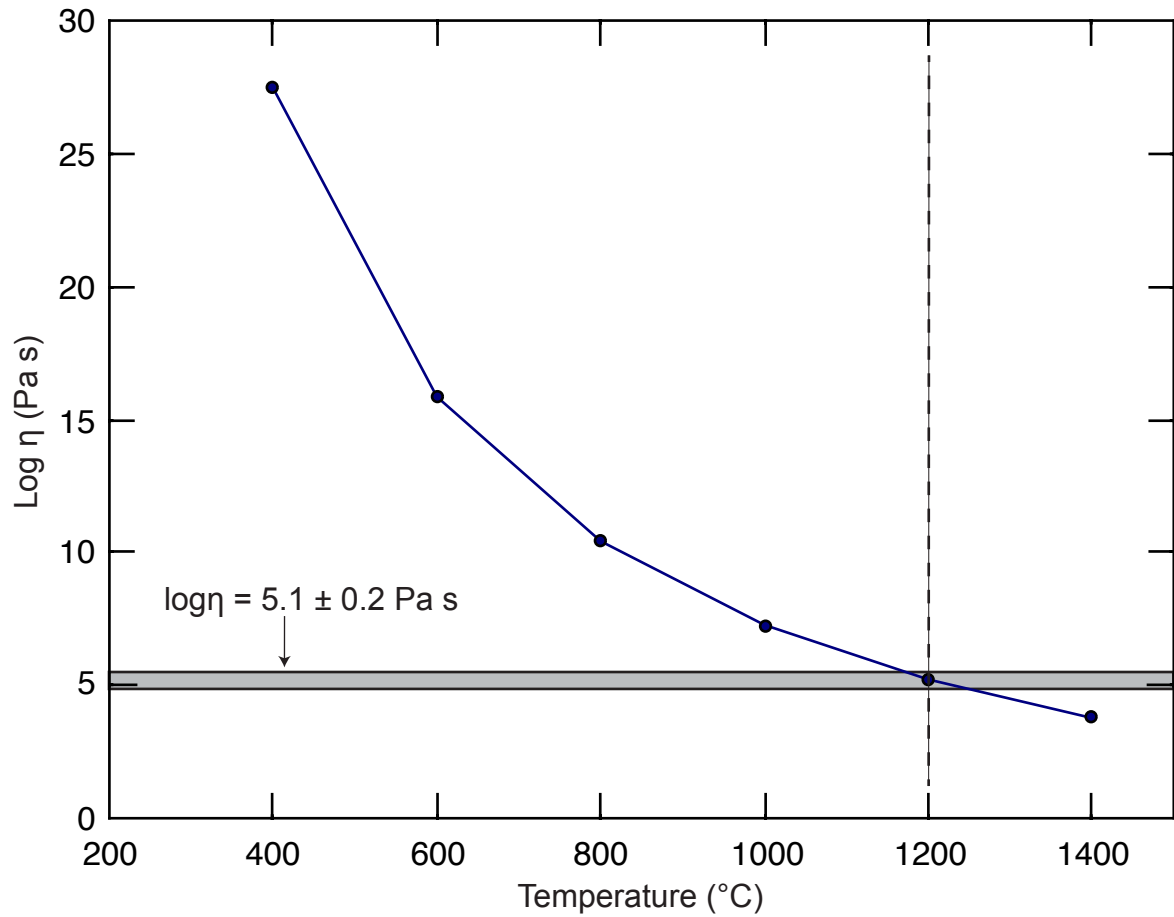
Figure 5

Figure 6

Latite



Rhyolite



—●— $\text{Log } \eta$ at 1200°C and 0 wt.% H_2O $\text{Log } \eta$ at 1000°C and 1.5 wt.% H_2O

Figure

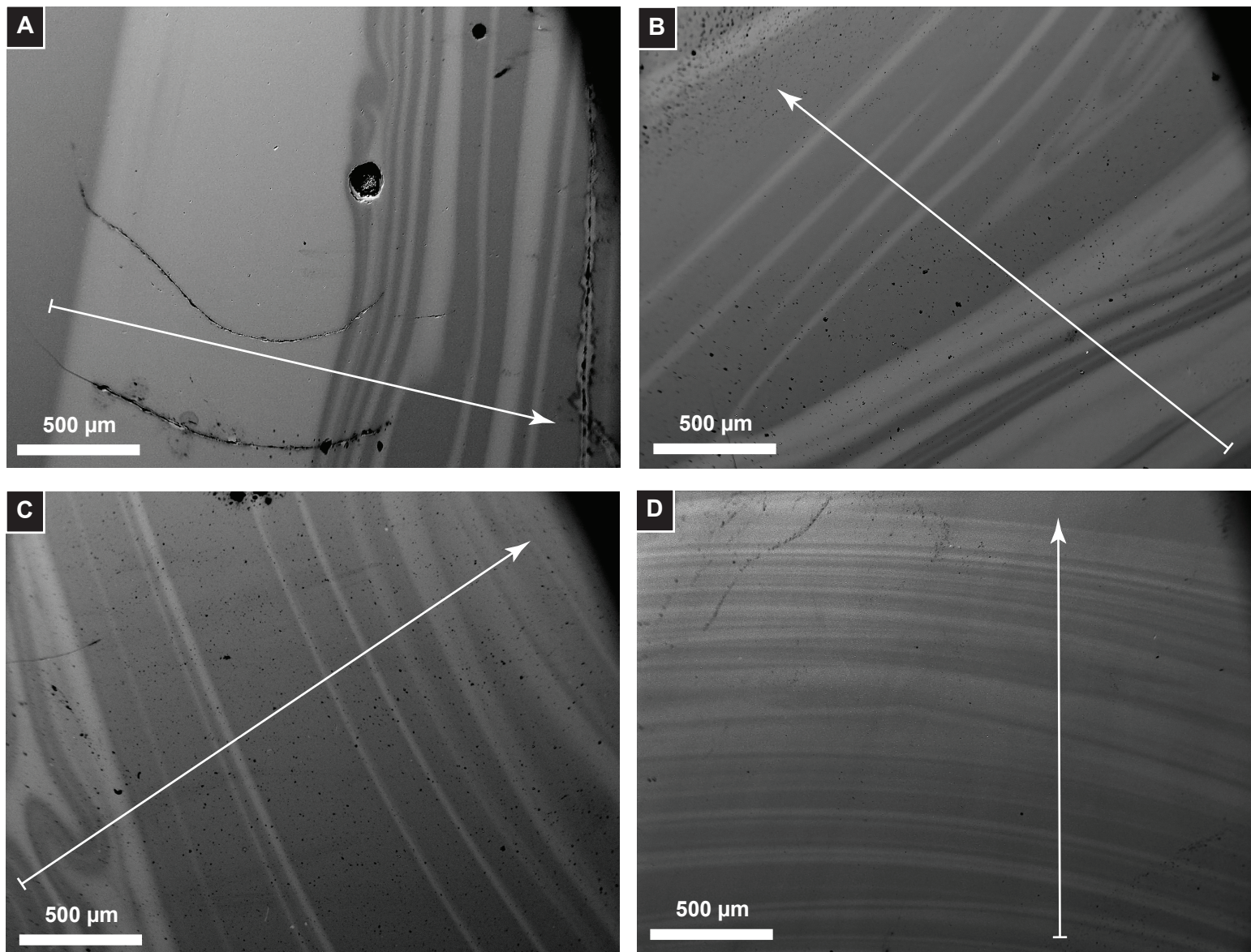


Figure 7

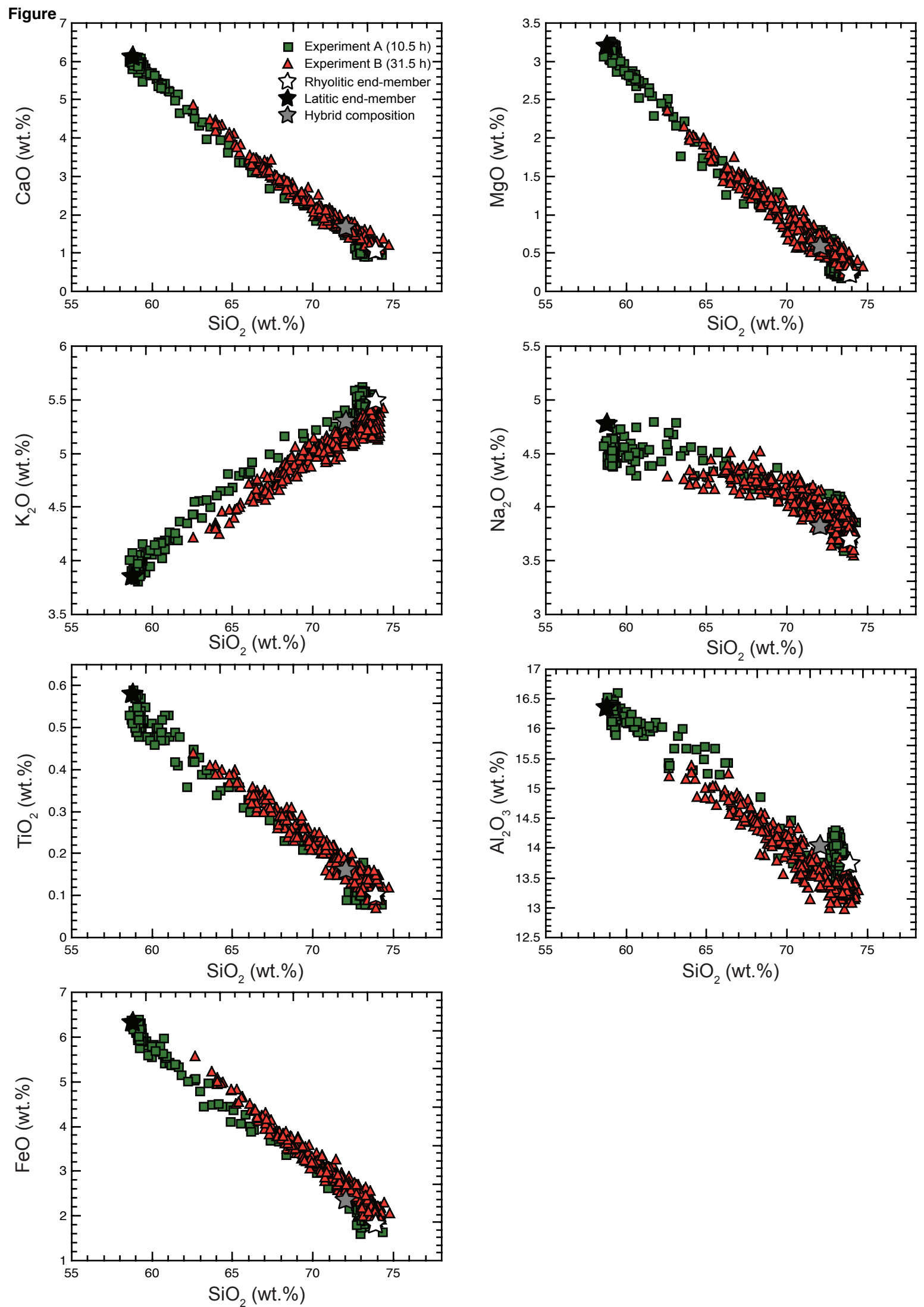


Figure 8

Figure

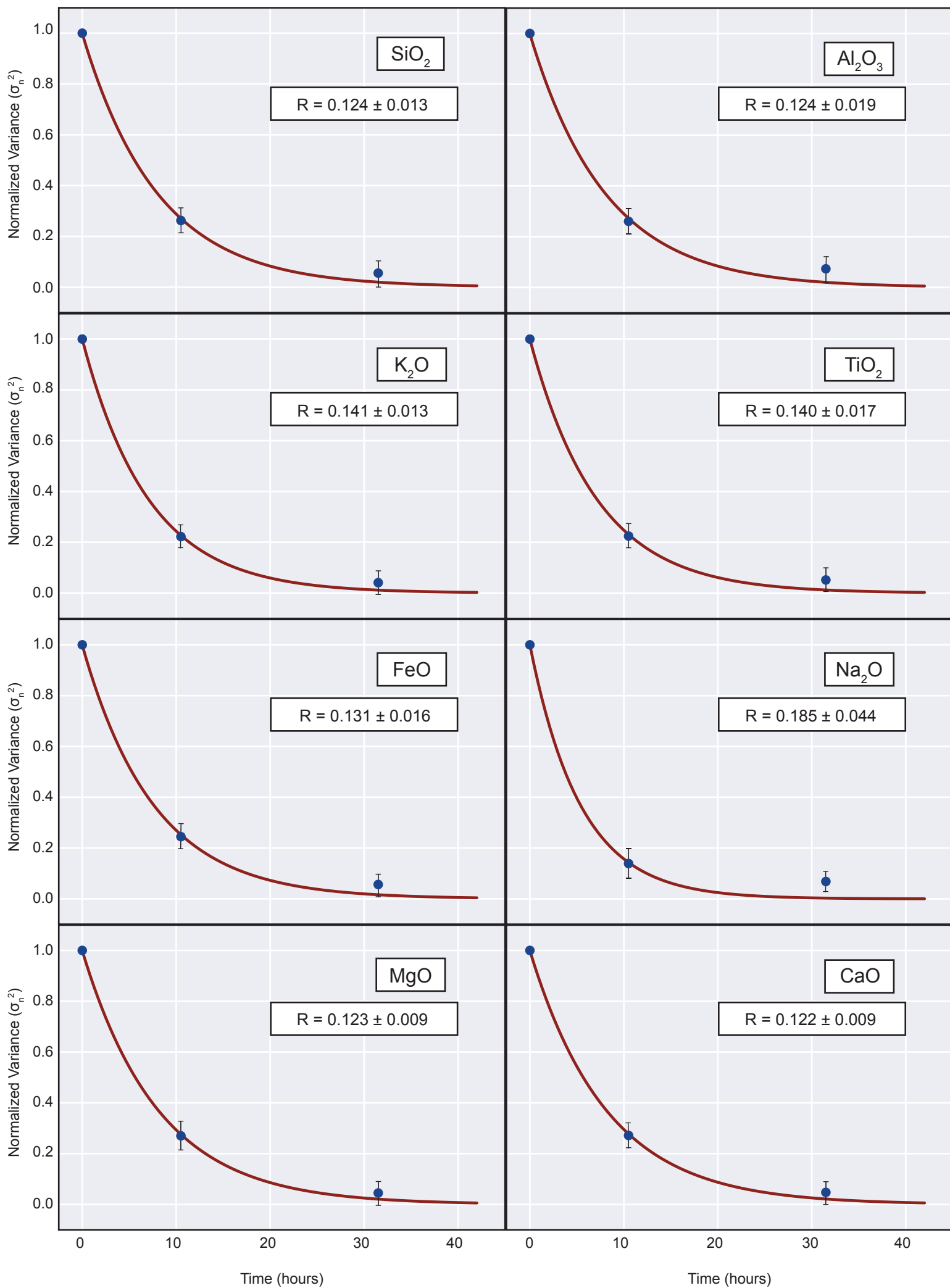
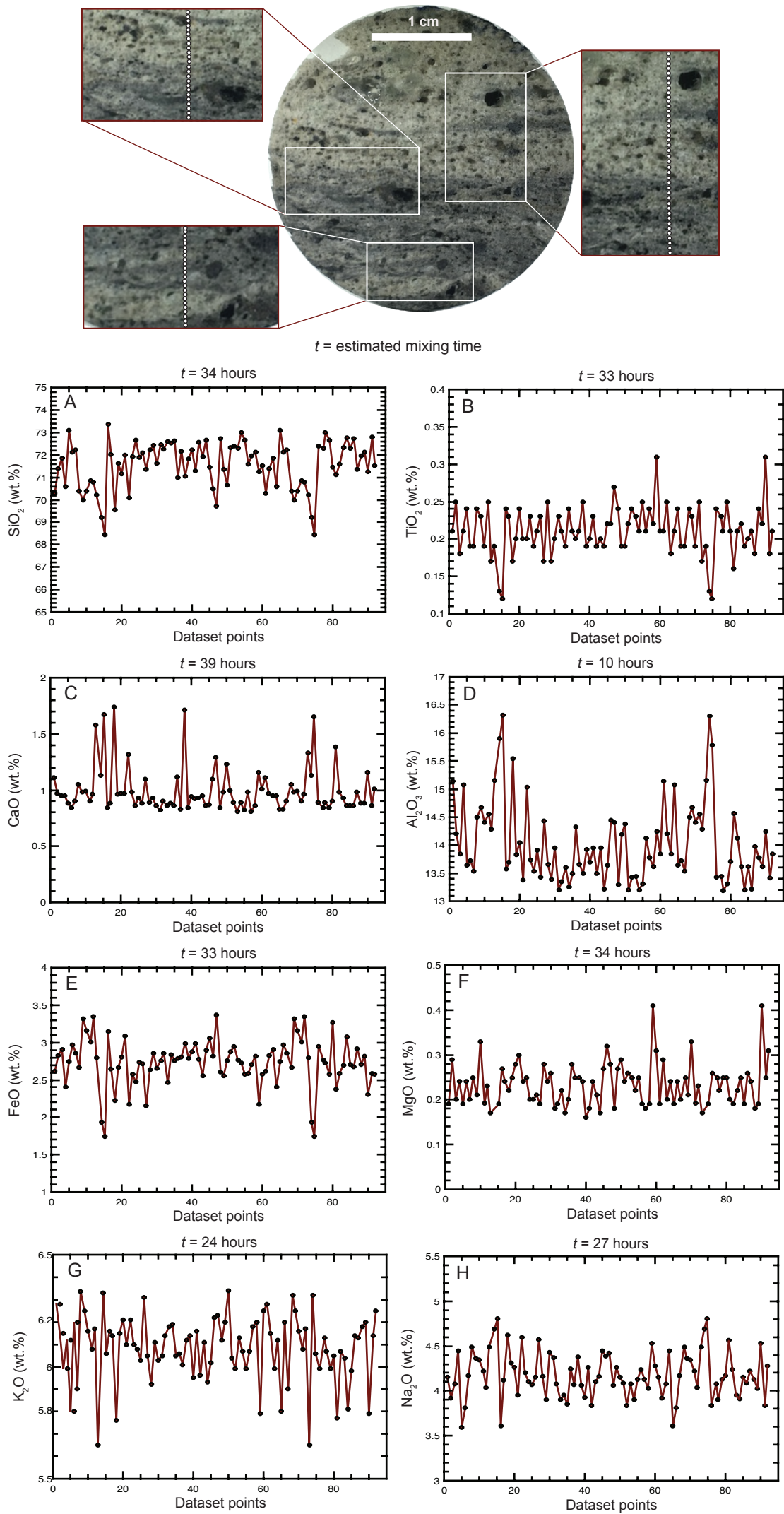


Figure 9



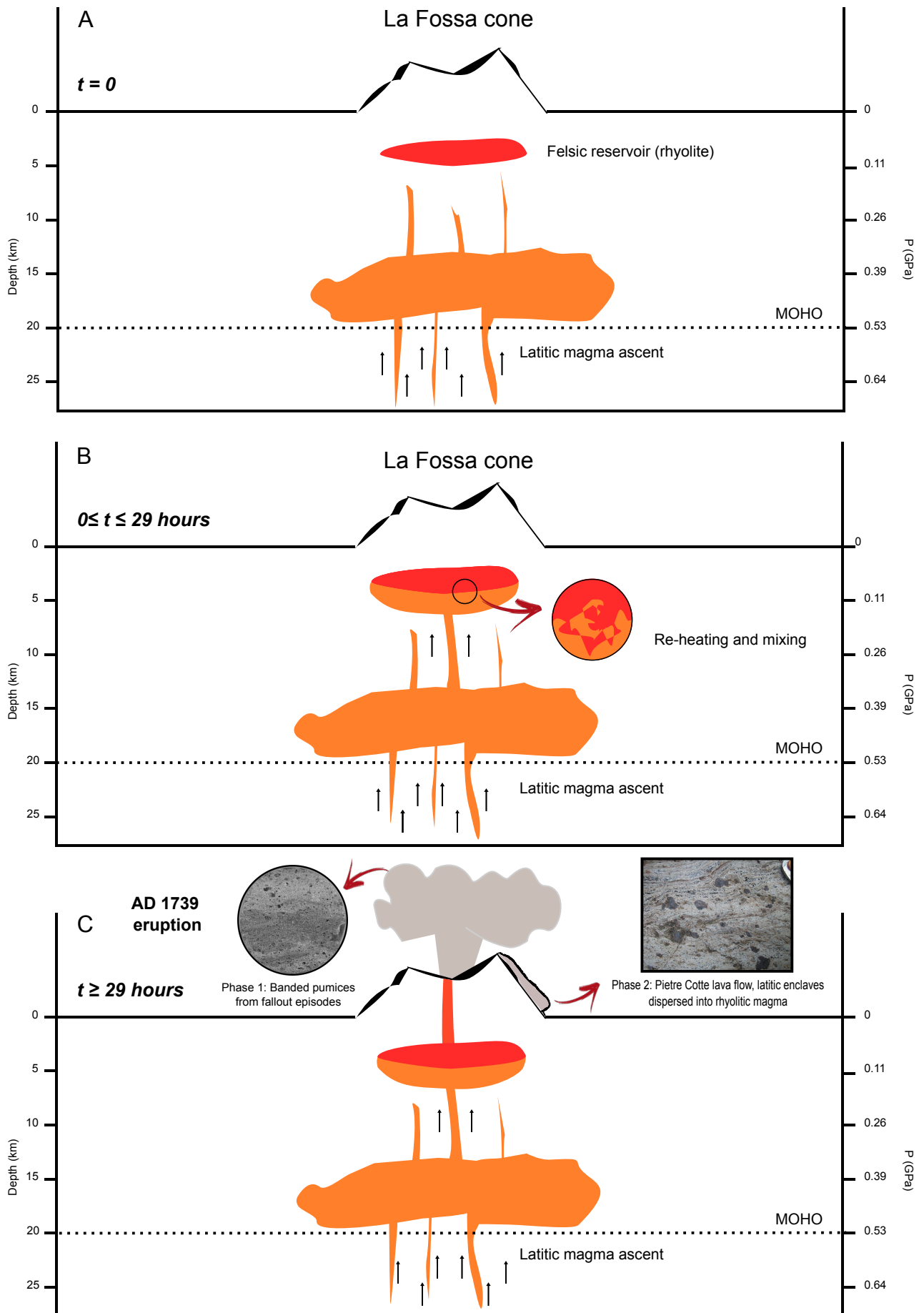


Figure 11

	Rhyolitic end-member (wt.%)	St. Dev.	Latitic end-member (wt.%)	St. Dev.
<i>N</i> =8				
SiO ₂	73.90	0.2	58.81	0.4
Al ₂ O ₃	13.73	0.3	16.36	0.2
K ₂ O	5.49	0.1	3.85	0.2
TiO ₂	0.10	0.1	0.58	0.1
FeO _t	1.81	0.1	6.31	0.1
Na ₂ O	3.69	0.3	4.77	0.2
MgO	0.22	0.1	3.20	0.1
CaO	1.05	0.1	6.12	0.1
Total	100.00		100.00	
Density (g/cm ³)	2.35		2.54	
Viscosity (Pa s) @1200°C	1.66 × 10 ⁵		1.62 × 10 ³	

Table 1

Table

[Click here to download Table: Table 2 .docx](#)

Island	Eruptive event	Evidence of magma mixing	References
VULCANO	AD 1739 Pietre Cotte eruption	Latitic enclaves dispersed in a rhyolitic host magma	<i>Perugini et al. (2007)</i> <i>Piochi et al. (2009)</i> <i>Vetere et al. (2015)</i>
SALINA	Porri lava flow, age ca. 43 ka Upper Pollara eruption, age ca. 13 ka	Basaltic-andesitic enclaves dispersed in a dacitic host magma Heterogeneous juvenile fragments (andesitic in composition) dispersed in a rhyolitic host magma	<i>De Rosa et al. (1996)</i> <i>Ventura et al. (2006)</i> <i>Perugini et al. (2004)</i>
LIPARI	Monte Guardia eruption, age ca. 22 ka Rocche Rosse lava flow, AD 1230 eruption	Latitic enclaves (showing zoned plagioclase and clinopyroxene) dispersed in a rhyolitic host magma Latitic enclaves dispersed in a rhyolitic host magma	<i>De Rosa et al. (2003)</i> <i>Gioncada et al. (2005)</i> <i>Davi et al. (2009, 2010)</i>
STROMBOLI	1985-2000 and 2002-2003 activity products 1985-1986, 1996 + Present-day activity products Post Pizzo pyroclastic sequence and Present-day activity products	Chemical zoning, resorptions and disequilibrium textures in clinopyroxene, olivine and plagioclase Radiogenic isotope variations in clinopyroxene, olivine and plagioclase Chemical zoning in clinopyroxene and plagioclase	<i>Landi et al. (2004, 2009)</i> <i>Armienti et al. (2007)</i> <i>Francalanci et al. (2012)</i> <i>Petrone et al. (2006, 2018)</i>

Table 2

Table

[Click here to download Table: Table SM1 .xlsx](#)

Dataset points	SiO ₂	Al ₂ O ₃	K ₂ O	TiO ₂
				Profile
1	72.86	14.23	5.19	0.10
2	72.87	14.12	5.24	0.14
3	73.09	14.09	5.25	0.13
4	73.17	14.06	5.22	0.13
5	72.97	14.04	5.22	0.12
6	73.46	13.83	5.24	0.11
7	72.82	14.26	5.23	0.10
8	73.16	14.27	5.26	0.12
9	73.28	14.08	5.24	0.12
10	73.17	13.94	5.25	0.10
11	73.17	14.03	5.19	0.13
12	73.19	13.99	5.28	0.09
13	73.15	14.19	5.24	0.13
14	73.16	14.35	5.33	0.10
15	73.10	14.07	5.22	0.11
16	73.27	14.02	5.30	0.14
17	73.14	14.00	5.30	0.10
18	73.26	13.92	5.32	0.10
19	73.26	14.15	5.30	0.13
20	73.05	14.10	5.32	0.12
21	73.02	14.10	5.25	0.09
22	73.18	14.01	5.34	0.11
23	73.14	14.25	5.25	0.09
24	72.83	14.27	5.28	0.13
25	73.24	14.07	5.29	0.11
26	73.12	14.10	5.26	0.13
27	73.43	13.96	5.31	0.13
28	72.76	14.18	5.37	0.11
29	73.28	14.06	5.32	0.13
30	73.32	14.01	5.25	0.12
31	73.04	14.13	5.27	0.14
32	73.20	14.24	5.36	0.11
33	73.15	13.95	5.35	0.16
34	73.11	13.97	5.33	0.11
35	73.01	14.07	5.25	0.09
36	73.02	13.91	5.48	0.08
37	73.20	13.90	5.46	0.10
38	73.49	13.89	5.24	0.13
39	73.17	14.09	5.32	0.12
40	73.26	14.00	5.33	0.12
41	72.93	14.13	5.40	0.12
42	73.59	13.78	5.42	0.12
43	73.15	13.77	5.43	0.14
44	70.35	13.75	5.10	0.21

Table[Click here to download Table: Table SM2.docx](#)

Oxide	C₁	C₀	R	<i>t</i> (hours)
SiO ₂	0	1	0.124	34
Al ₂ O ₃	0	1	0.124	10
K ₂ O	0	1	0.141	24
TiO ₂	0	1	0.14	33
FeO	0	1	0.131	33
Na ₂ O	0	1	0.185	27
MgO	0	1	0.123	34
CaO	0	1	0.122	39
<hr/>				
averaged <i>t</i>				29
stand. dev.				9

Table SM2